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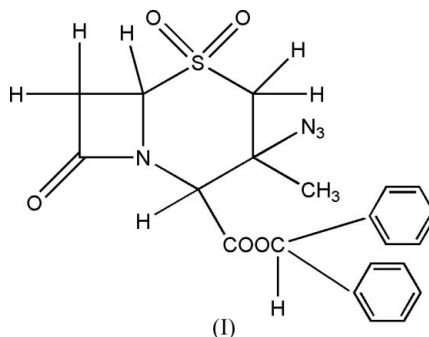
Key indicators

Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.042
 wR factor = 0.091
Data-to-parameter ratio = 13.4For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

Diphenylmethyl 3-azido-1,1-dioxocephalosporanate

The title compound, $\text{C}_{21}\text{H}_{20}\text{N}_4\text{O}_5\text{S}$, crystallizes with two
molecules in the asymmetric unit. The crystal structure is
stabilized by a network of $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds.Received 14 November 2006
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Comment

Tazobactam is a widely used beta-lactamase inhibitor (Bai *et al.*, 2001, Micetich *et al.*, 1987). The title compound, (I), is a by-product of the synthesis of tazobactam and the structure of its benzene solvate has been reported (Liu, 2006). The unsolvated material, (I), was obtained from 6-aminopenicillanic acid, and its structure is reported here (Figs. 1 and 2).Compound (I) crystallizes with two independent, but
structurally quite similar, molecules in the asymmetric unit.
All bond lengths and angles in (I) are within normal ranges
(Allen *et al.*, 1987) and similar to those reported for the
solvated material (Liu, 2006). The four-membered azetedi-
none rings are planar (r.m.s. deviations 0.0246 and 0.0249 Å).
The thiazine rings adopt chair conformations. The C1/O1/O2/
C14/C15 and C22/O6/O7/C35/C36 carboxylate units are also
planar (r.m.s. deviations 0.0232 and 0.0383 Å, respectively)
and lie approximately orthogonal to the respective thiazine
rings. The crystal structure is stabilized by a network of
intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds (Fig. 3 and Table 1).

Experimental

The title compound was prepared by the procedure of Bai *et al.*
(2001). Colourless single crystals of (I) were grown by slow
evaporation of a methanol solution.

Crystal data

$\text{C}_{21}\text{H}_{20}\text{N}_4\text{O}_5\text{S}$	$Z = 8$
$M_r = 440.47$	$D_x = 1.375$ Mg m ⁻³
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
$a = 10.9030$ (13) Å	$\mu = 0.19$ mm ⁻¹
$b = 11.3796$ (14) Å	$T = 294$ (2) K
$c = 34.293$ (4) Å	Block, colourless
$V = 4254.8$ (9) Å ³	$0.24 \times 0.22 \times 0.18$ mm

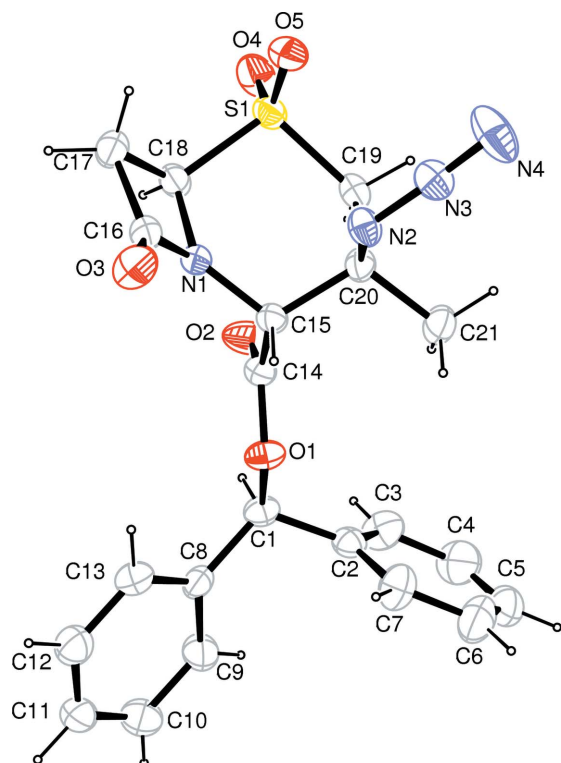


Figure 1

One of the two molecules (molecule 1) in the asymmetric unit of (I), with the atom-numbering scheme and 30% probability displacement ellipsoids.

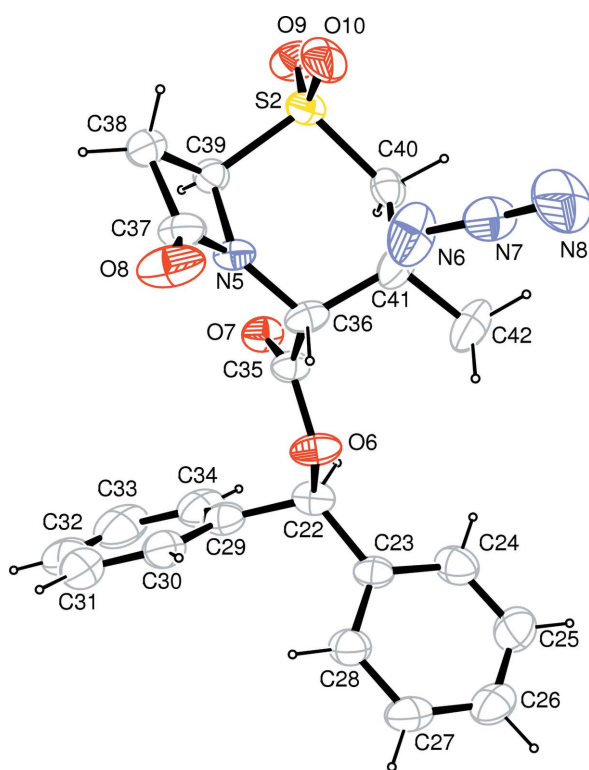


Figure 2

The other molecule (molecule 2) in the asymmetric unit of (I), with the atom-numbering scheme and 30% probability displacement ellipsoids.

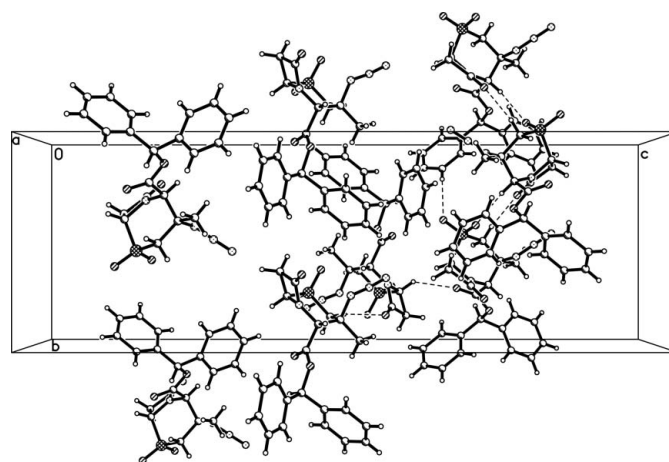


Figure 3

Packing diagram for (I), with hydrogen bonds drawn as dashed lines.

Data collection

Bruker SMART-1000 CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.943$, $T_{\max} = 0.966$

20301 measured reflections
 7497 independent reflections
 5529 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.037$
 $\theta_{\max} = 25.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.092$
 $S = 1.03$
 7497 reflections
 561 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0408P)^2 + 0.1929P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.20 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.25 \text{ e } \text{\AA}^{-3}$
 Absolute structure: Flack (1983),
 2607 Friedel pairs
 Flack parameter: 0.04 (6)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C19—H19B \cdots O3 ⁱ	0.97	2.44	3.364 (3)	160
C31—H31 \cdots O9 ⁱⁱ	0.93	2.57	3.446 (6)	157
C36—H36 \cdots O10 ⁱⁱⁱ	0.98	2.25	3.196 (4)	162
C40—H40B \cdots O8 ^{iv}	0.97	2.29	3.235 (4)	166
C18—H18 \cdots O7 ⁱⁱ	0.98	2.50	3.262 (3)	135

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{5}{2}, -z + 2$; (ii) $x, y + 1, z$; (iii) $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$; (iv) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$.

All H atoms were positioned geometrically and refined using a riding model, with $C-H = 0.93 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for aromatic, $C-H = 0.98 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for CH, $C-H = 0.97 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ for CH_2 , and $C-H = 0.96 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for CH_3 H atoms.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

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References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bai, G. Y., Chen, L. G., Li, Y. & Cao, L. (2001). *Fine Chem.* **18**, 634–637.
- Bruker (1997). *SMART* (Version 5.01), *SAINT* (Version 5.01) and *SHELXTL* (Version 6.1). Bruker AXS Inc., Madison, Wisconsin, USA.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Liu, X.-L. (2006). *Acta Cryst.* **E62**, o4431–o4432.
- Micetich, R. G., Maiti, S. N., Spevak, P., Hall, T. W., Yamabe, S., Ishida N., Tanaka, M., Yamazaki, T., Nakai A. & Ogawa, K. (1987). *J. Med. Chem.* **30**, 1469–1474.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.